

Production and characterization of chitinase from *Vibrio* species, a head waste of shrimp *Metapenaeus dobsonii* (Miers, 1878) and chitin of *Sepiella inermis* Orbigny, 1848

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ABSTRACT

The chitin is extracted from the cuttlebone of *S. inermis* and the mineral contents are predictable. The structure and degree of deacetylation of extracted cuttlebone chitin is dogged through Fourier Transform Infrared (FT-IR) spectroscopy. The extracted cuttlebone chitin is used as the substrate for the production of chitinase from *Vibrio* sp. The extra cellular proteins are concentrated by ammonium sulphate precipitation, dialysed and then purified by using gel (sephadex G-100) chromatography. Among the 40 fractions, only two fractions (active fractions) showed maximum absorbance at 280 nm, which are pooled, dialysed and free-dried. The enzyme activity (0.104 μ moles/ml) and molecular weight (50 - 60 kDa) of purified chitinase is also determined through SDS-PAGE. The optimal condition for chitinase activity is pH between 6.0 - 6.5 and 45°C.

Keywords: Chitin; *S. inermis*; Chitinase; *M. dobsonii*; *Vibrio* sp.

1. INTRODUCTION

Chitin is one of the most abundant biopolymers in nature and possibly the most abundant in the marine environment, since it is produced by many marine organisms, including zooplankton and several phytoplankton species [1]. The other popular source is the mycelia (containing up to 20% chitin) from mushrooms and other fungi. About an average of 30% and up to 60% of the cuticles of insects are chitins [2]. Chitin production is estimated approximately 1×10^{11} tons annually and it is the second most abundant natural biopolymer. Because chitin has many useful biological properties such as biocompatibility, biodegradability, haemostatic activity and wound

healing property, much attention has been paid to its bio-medical applications; e.g. an absorbable suture, a drug carrier, an antitumor agent [3].

Chitin is composed of repeating *N*-acetyl D-glucosamine residues and promising biopolymer with numerous industrial, medical and commercial uses. However, it is difficult to purify and modify chemically. Hence identification of chitin modifying enzymes and elucidation of their activities could facilitate the efficient production of specific chitin products. The biodegradation of chitin requires the synergistic action of several hydrolytic enzymes for efficient and complete breakdown. The combined action of endo and exochitinases results in the degradation of chitin polymer into the soluble *N*-acetyl D-glucosamine [4].

Chitinases are hydrolyzing the chitin; these enzymes have been implicated in a wide range of biological events including host infection, growth and development, defense against pathogens and human diseases. The research of chitinases is most important, because of its biological controller in ecology, epidemiology of diseases and the preparation of medicines. In the marine environment, most chitin seems to be degraded by marine bacteria. Some chitinolytic bacteria, including *Aeromonas*, *Pseudomonas*, *Clostridium* and *Vibrio* have been isolated and their chitinase activity was also studied [5]. Hence an attempt has been made to isolate chitin from *S. inermis* and its structure is determined through FT-IR spectrum which is used for the production and characterization of chitinase from *Vibrio* sp.

2. MATERIALS AND METHODS

2.1. Extraction of Chitin

Chitin is prepared by following the method of Takiguchi [6].

2.2. Estimation of Minerals

The demineralized water of cuttlebone chitin is incubated

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with concentrated HNO₃ for 24 hr after that mixture of HNO₃: Perchloric acid mixture (4:1) is added and incubated. Then digested at 80°C to the obtained white precipitate 10% (v/v) HNO₃ is added and filtered. The total volume is made up to 20 ml and the minerals are estimated through ICP [7].

2.3. FT-IR Spectral Analysis

The quality parameter of chitin is analyzed by FT-IR spectroscopy [8]. FT-IR spectrum of solid samples of chitin is relied on a Bio-Rad FTS-40 model, USA and its average degree of acetylation is calculated by following the method of Baxter *et al.* [9].

2.4. Isolation of Microorganism

Heads of soft shell shrimps (*M. dobsonii*) are collected from the Cuddalore landing center (Lat.11°24'N; Long 79°49'E), Tamil Nadu, India. Samples are homogenized and diluted 10⁻¹ to 10⁻⁵ in sterile distilled water and spread on chitin—containing minimal agar plates. After incubation for 2 weeks at room temperature, clear zone forming bacteria are selected as chitinase producer [5].

2.5. Production of Chitinase Enzyme

For the production of chitinase from *Vibrio* sp. is grown in 100 ml of medium consisting of 0.35% K₂HPO₄, 0.15% KH₂PO₄, 0.1% MgSO₄·7H₂O, 0.1% tryptone, 0.1% (w/v) yeast extract, and 1% (w/v) extracted cuttlebone chitin powder in a 250 ml Erlenmeyer flasks for a further culture at 28°C for 4 days. Cells induced by using powdered chitin are harvested by centrifuged at 12,000 g for 20 min at 4°C. The supernatant is precipitated by adding 90% (w/v) ammonia sulfate and then dialyzed against 50 mM sodium phosphate buffer (pH 6.0) to remove salts and the crude enzyme was thus obtained [10].

2.6. Preparation of Colloidal Chitin

Colloidal cuttlebone chitin is prepared by the method of Rodriguez-Kabana *et al.* [11] by partial hydrolysis of chitin with 10 N HCl for 2 hr at room temperature. The colloidal chitin is washed several times with large volumes of distilled water to adjust the pH to 7.0.

2.7. Assay of Chitinase Activity

Chitinase activity is assayed by detecting the reducing groups produced by the chitinase reaction using colloidal extracted cuttlebone chitin as the substrate [11]. The assay is performed according to the procedure previously reported [12]. One unit (U) of chitinase activity is defined as the amount of enzyme that produced 1 μmol of

reducing equivalent per minutes at 37°C. The monomer of chitin, GlcNAc, is taken as the standard for the reduce sugar [13]. Briefly, a 0.13 ml enzyme preparation is incubated with 0.87 ml of 0.2% (w/v) colloidal chitin azure solution at 37°C for 30 min. The reaction mixture is then centrifuged and the absorbance of the supernatant is measured at 550 nm.

2.8. Effect of pH and Temperature Optima on the Activity

The effect of pH on the purified chitinase is determined at different pH values (4.5 to 10.0) under standard assay conditions. The buffer systems used are as follows: 0.1 M citrate-phosphate buffer, pH 4.5 - 6.5; 0.1 M phosphate buffer, pH 6.5 - 8.5; 0.1 M glycine-NaOH buffer, pH 8.5 - 10.0. The optimal temperature for the chitinase activity is determined in the range of 20°C to 50°C under standard assay conditions. The purified enzyme is preincubated for 5 min at the temperature tested. The reaction was started with the addition of para nitrophenyl β-D-acetylglucosaminide.

2.9. Estimation of Protein

Protein concentrations are measured according to Bradford's [14] method using BSA as the standard.

2.10. Purification and Molecular Weight Determination of Chitinase

After production, the cells are removed by centrifugation at 5000 × g and 4°C for 20 min. Protein in the cell-free culture broth (900 ml) are precipitated with ammonium sulfate (30%, w/v). The precipitate is obtained by centrifugation (16,000 × g, 30 min, 4°C) and suspended in 3 ml of 0.1 M citrate-phosphate buffer (pH 6.0). The suspension is eluted through sephadex G-100 (Sigma) in a glassy column (100 × 1.0 cm). The elution buffer is 50 mM Tris-HCl and the volumetric flow rate is 0.33 ml/min. Every 4 ml from the effluent of gel filtration is collected in a fraction [15]. SDS-PAGE analysis of proteins is carried out by using 12.5% (w/v) polyacrylamide gel at 120 V for 90 min. Solutions for preparing 12.5% (w/v) resolving gel included H₂O (6.4 ml), 30% acrylamide mix (8.3 ml), 1.5 M Tris (pH 8.8, 5 ml), 10% (w/v) SDS (0.2 ml), 10% (w/v) ammonium persulfate (0.1 ml), and *N,N,N,N*-tetramethylethylenediamine (0.008 ml). The gel separation was performed based on a suggested protocol [16] except for eliminating the step of heating sample to 100°C for 3 min prior to electrophoresis.

3. RESULTS AND DISCUSSION

The percentage yield of chitin was found to be 22%.

3.1. Mineral Study

Minerals such as Ca (5.90%), P (0.05%), Cu (0.002%), Fe (0.5%), Mg (0.04%) and Zn (0.07%) of cuttlebone of *S. inermis* were recorded.

3.2. FT-IR Spectral Analysis

The FT-IR spectrum of the standard and cuttlebone chitin was obtained and the effective peaks were compared with the standard chitin (**Table 1**).

The FT-IR spectrum of the standard chitin showed twelve major peaks at the range of 897.41, 1026.63, 1077.93, 1154.64, 1259.54, 1422.73, 1587.94, 1660.55, 2361.41, 2922.85, 2922.85 and 3377.95 cm^{-1} ; whereas the cuttlebone chitin from *S. inermis* recorded the peaks in the range of 470.40 to 3753.13 cm^{-1} (Figures not shown) [17].

3.3. Degree of Deacetylation

The degree of deacetylation was calculated as 55.95%.

3.4. Isolation of Microorganism

On the chitin agar clear zone producing organism was selected for further chitinase study and the organism was identified as *Vibrio* sp. (**Figure 1**) through the biochemical test. Golden yellow colonies were produced on the TCBS agar [18].

3.5. Production of Chitinases

Vibrio sp. was grown aerobically in 100 ml of the opti-

imum medium in a 250 ml Erlenmeyer flask at 30°C. During the process of incubation, cell growth (as judged by OD₆₆₀) and pH were measured. Consequently it was inferred that chitinase enzymes produced by *Vibrio* sp.

3.6. Purification of Enzyme

Extra cellular chitinase was partially purified by gel chromatography using sephadex G-100 as column, 40 fractions were collected. Among the 40 fractions two fractions (6 and 14) showed maximum absorbances (**Figure 2**), which were pooled and used for further studies.

3.7. Activity of Chitinase

The conventional method using colloidal chitin as the substrate is based on the evaluation of reducing sugars released due to the action of chitinase. Incubation of GlcNAc with the colour reagent solution that was made by dissolving 0.5 g potassium ferricyanide in 1 lit of 0.5 M sodium carbonate led to the decrease in the absorbance at 420 nm, due to the reducing power of GlcNAc. For the determination of enzymatic activity, a calibration curve up to GlcNAc concentration of 0.104 $\mu\text{mol/ml}$ showing the release of reduce sugar (in 30 min) from the cleavage of chitin versus the chitinase concentration was thus obtained as shown in **Figure 3**.

3.8. Determination of Molecular Weight

The molecular weight of purified chitinase was found to be 50 - 60 kDa. It was confirmed through SDS-PAGE by compared with protein molecular marker (Genei) (20 -

Table 1. FT-IR spectrum of standard and cuttlebone chitin of *S. inermis*.

Regions	Standard chitin (cm^{-1})	Cuttlebone chitin (cm^{-1})
OH out-of-plane bending	690	613
NH out-of-plane bending	752	710
Ring stretching	896	860
CH ₃ wagging along chain	952	-
CO stretching	1026	1028
CO stretching	1073	1080
Asymmetric in-phase ring stretching mode	1116	1116
CH ₂ bending and CH ₃ deformation	1418	1477
Amide II band	1563	-
Amide I band	1661	1656
CH stretching	2878	2853
Symmetric CH ₃ stretching and asymmetric CH ₂ stretching	2930	2918
NH stretching	3268	-
OH stretching	3436	3413

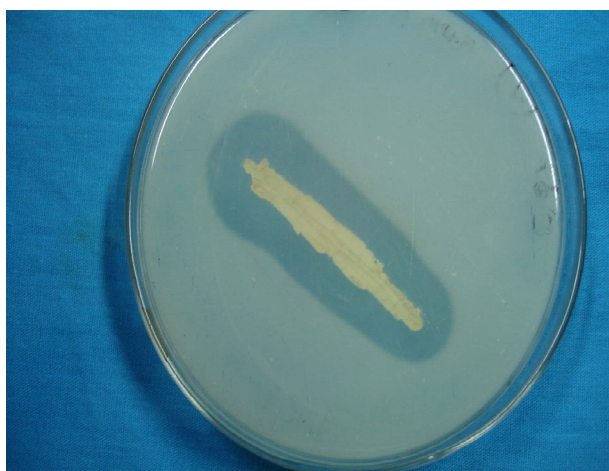


Figure 1. *Vibrio* sp. in TCBS agar medium.

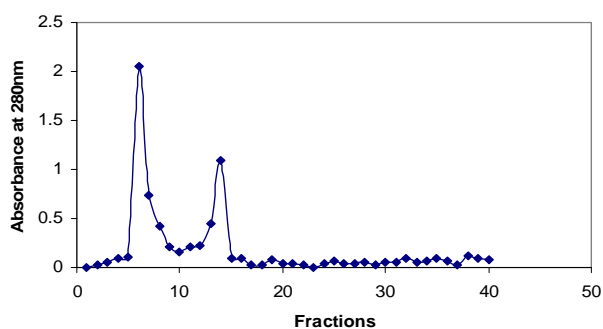


Figure 2. Gel chromatography of chitinase.

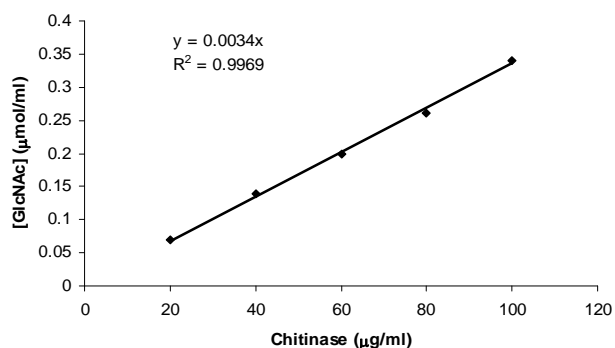


Figure 3. Determination of chitinase activity by the method of standard reducing sugar.

200 kDa) and chitinase (Sigma) (**Figure 4**).

3.9. pH Optimum of Chitinase

The effect of pH of the chitinase was studied with citrate-phosphate buffer (pH 4.5 - 6.5), phosphate buffer (pH 6.5 - 8.5), and glycine-NaOH buffer (pH 8.5 - 10.0) under the standard conditions. The pH activity of chitinase was maximum between pH 6.0 - 6.5 (**Figure 5**). Similarly the other marine bacterial chitinases showed broader pH optima or were more active in neutral or slightly alkaline

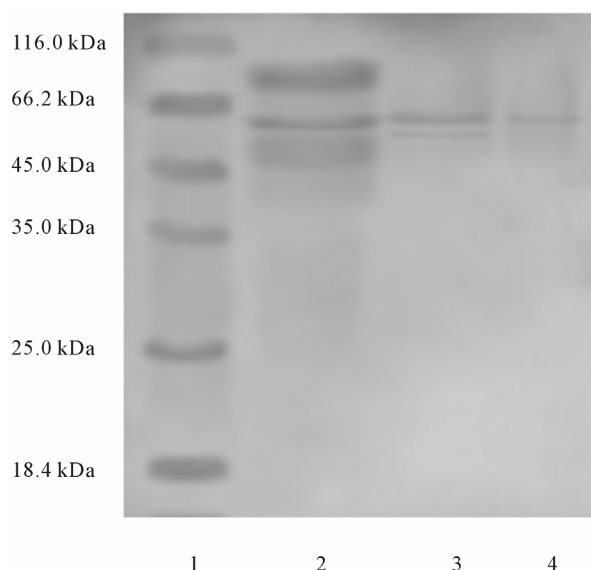


Figure 4. SDS-PAGE of chitinase. Lane 1: Molecular weight markers; Lane 2: Ammonium sulphate precipitated proteins; Lane 3: Sephadex G-100 purified chitinase; Lane 4: Standard chitinase.

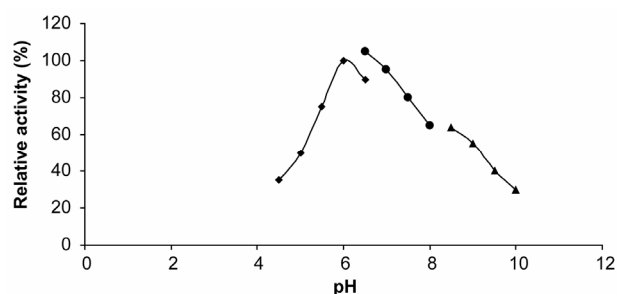


Figure 5. Effect of pH on activity of chitinase (■: citrate-phosphate buffer; ●: phosphate buffer; ▲: glycine-NaOH buffer).

conditions [19].

3.10. Effect of Temperature of Chitinase

The optimum temperature of chitinase was 45°C (**Figure 6**). Chitinase I of strain 98CJ11027 showed a similar temperature optimum to those of other marine bacteria (*Aeromonas hydrophila* H-2330) [20], *Alteromonas* sp. and *Pseudomonas aeruginosa* K-187 [19].

In the present investigation chitin extracted from the cuttlebone of *S. inermis* was found to be 22%, this is higher than the result of Tolaimate *et al.* [7], reported 20% of chitin was found in the cuttlebone of *S. officinalis*. Whereas, in general, the squid/ octopus contain 3% - 20% and dry *Squilla* contains 15% of chitin [21].

Zhou *et al.* [22] found that the chitinase enzyme showed optimum activity pH at 6.5 and optimum temperature at 50°C. Further it was stable in the pH range of 4 to 9 and temperature below 40°C, during 60 and 120 min treat-

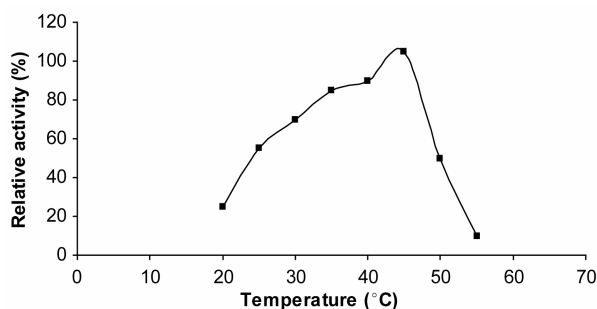


Figure 6. Effect of temperature on chitinase activity.

ment respectively. The enzyme produced by *Vibrio* sp. was found to be represented by a single protein band of molecular weight about 30 kDa, on SDS-PAGE. In terms of molecular mass comparison, the chitinase from *Vibrio* sp. 1211 seems to be different from those of other microbes. Almost all of the chitinases from microorganisms have higher molecular weights: 42 kDa for the enzyme from fungi *Piromyces* sp.; 59 - 89 kDa for that from *Bacillus lichiniiformis*; 41 - 44 kDa for that from *Streptomyces* sp. The chitinases from the marine bacterium *Aeromonas hydrophila* also reported a higher molecular weight of 62 kDa. The present study, the pH (6.0 - 6.5), temperature (45°C) and the molecular weight of the chitinase produced by *Vibrio* sp. was found to be higher and it was estimated as 50 - 60 kDa.

Dahiya *et al.* [4] observed that *Enterobacter* sp. NRG4 secretes extracellular chitinase and molecular weight was found to be about 60 kDa. Chitinase from *Enterobacter* sp. NRG4 was active over a broad pH range *i.e.* from pH 4.5 to 8.0, optimum being 5.5. Wang *et al.* [10] found that the chitinase activity increased along with the cell growth (as judged by OD₆₆₀) and reached maximum when the cell growth reached mid-exponential phase at 3 days of incubation and the molecular mass of *Bacillus subtilis* W-118 chitinase was estimated as 20.6 kDa. Park *et al.* [5] isolated twenty eight different chitinase producing bacteria and they identified through the morphological, physiological, biochemical characteristics and 16s rDNA sequence as all are member of the genus *Vibrio*.

4. CONCLUSION

The chitin yield was 22% in the cuttlebone of *S. inermis* and the molecular weight of the purified extra cellular chitinase was found to be 50 - 60 kDa, optimum pH between 6.0 - 6.5 and the optimum temperature was found to be 45°C. Further study in this line is required to purify the enzyme by HPLC, sequenced, determine the kinetics and synthesized by rDNA technology.

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